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Influence of Cell Size on Performance of Lithium Ion Battery

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Motivation

- > Experimental
- Simulation Model
- Results and Discussion
- Conclusion and Outlook







- Large format cells are becoming population in EV Applications
- Lesser connections and fewer cells to monitor for BMS unit
- Different designs and approaches to produce large format cells
- Problems
 - Safety challenges due to high energy content
 - Temperature non-uniformities
 - Inhomogeneous current density and SOC distribution
- > Consequence
 - Accelerated localized degradation of power performance
 - Lifetime reduction
 - Effects on battery safety
 - Different Concepts of thermal management



Experimental (1/4)



Nominal Capacity (Ah)	Size (mm) (WxHxT)	Weight (g)
8	105x100x7.05	157
25	225x224x6	570
53	225x224x12.3	1,200
75	263x266x11.2	1,500

Cathode: Nickel Manganese Cobalt Oxide Anode: Graphite Nominal Voltage: 3.7 V Electrolyte: LiPF₆ with EC:EMC





Experimental (2/4)



[1] K. Lee, "Voltage Relaxation of LiyFePO4 based Batteries," Master Thesis, Technical University Munich, Munich, 2014





Experimental (3/4)

HPPC-Measurements:

- 11 Measurements over SOC
- 15°C, 25°C, 40°C
- Short time transient resistance
- Long time long time resistance

EIS Measurement:

- 14 Measurements over SOC
- 15°C, 25°C, 40°C, 50°C
- 6kHz 10mHz
- Internal resistance



 t_2



Test profile for 25 Ah cell at 25°C

R

Current



Experimental (4/4)





Command	Parameter	Limit
Charge	I = 1CA U = 4.2 V	I< 0.05 CA
Pause	3 h	
Discharge	l = 0.5, 1, 2, 3 C U= 2.7 V	l < 0.05 CA
Temperature	T = 15, 25 & 40°C	

25/53 Ah Cell









Simulation Model(1/5)

- >Scale up approach
- > Microscopic parameters are same
- Surface area, tab positions and size are different
- Scale modelling approach with coupling
 Simulation to study temperature and voltage





Simulation Model(2/5)

Electrochemical Model



Basics :

- Porous Electrode Theory
- Concentrated Solution Theory
- Ohm's Law
- Fick's Law
- Butler-Volmer Equation

Cathode

Anode



- 9 electrochemical sub-models
- Current collector segmented into 9 pieces
- Model non-uniform electrochemical performance





Simulation Model (3/5)



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10



Simulation Model (4/5)

Schematic of coupling of sub models







Simulation Model (5/5)

Term	Equations
Transport in solid phase	$\frac{\partial c_s}{\partial t} = \frac{1}{r^2} \frac{\partial}{\partial r} (D_s r^2 \frac{\partial c_s}{\partial r})$
Transport in electrolyte	$\varepsilon \frac{\partial c_l}{\partial t} = \nabla \cdot \varepsilon D \nabla c_l - \frac{i_l \cdot \nabla t^0_+}{F} + a j_n (1 - t^0_+)$
Potential in Solid Phase	$i_s = -\sigma_{eff} abla \phi_s$
Potential in Electrolyte Phase	$\nabla \Phi_l = -\frac{i_l}{\kappa} + \frac{2RT}{F} (1 - t_+^0) \left(1 + \frac{d\ln f_{\pm}}{d\ln c_l} \right) \nabla \ln c_l$
Electrochemical Kinetics	$j_n = i_0 \left[\exp\left(\frac{\alpha_a F}{RT}\eta\right) - \exp\left(\frac{\alpha_c F}{RT}\eta\right) \right]$ $\eta = \Phi_s - \Phi_l - U - F \cdot j_n \cdot R_{film}$ $i_0 = F(k_a)^{a_c} (k_c)^{a_a} (c_{s,max} - c_s)^{a_c} (c_s)^{a_a} (c)^{a_a}$
Equilibrium Potential	$E_{eq} = E_{eq,ref} + (T - T_{ref}) * \frac{dU}{dT}$
Temperature distribution	$\rho C_p \frac{dT}{dt} = \nabla (k\nabla T) + Q_{irr} + Q_{conv} + Q_{rev}$ $Q_{irr} = Q_{ohm} + Q_{pol}$ $Q_{ohm} = \sigma_{eff} \nabla \varphi_s \nabla \varphi_l + \sigma_{eff} \nabla \varphi_l \nabla \varphi_l + \kappa_{eff} \nabla \ln c_l \nabla \Phi_l$ $Q_{conv} = h_{conv} (T - T_{ref})$ $Q_{rev} = I * T * \frac{\partial E_{eq}}{\partial T}$





Results & Discussion (1/10)

Terminal Voltage and Temperature Validation at 40°C







- Simulated Voltage is in good agreement with measured data
- At low discharge rate simulated voltage under estimates the discharge capacity
- Temperature rise at the beginning of discharge is high during simulation





Results & Discussion (3/10)

Max. Temperature Gradient (ΔT_{MAX}) with forced air circulation







Results & Discussion (4/10)

Max. Temperature Gradient (ΔT_{MAX}) with Al. Cooling plates







> ∆T_{MAX} is highest at low ambient temperature

- $ightarrow \Delta T_{MAX}$ increases with increase in cell size, with the exception for 25 Ah cell





Results & Discussion (6/10)

Internal Resistance of Cells measured from Electrochemical Impedance Spectroscopy







Results & Discussion (7/10)

➤ R_i is sensitive to temperature change. Hence ΔT is highest at low ambient temperature

It does not increase significantly with SOC decrease, except near end of discharge





Results & Discussion (8/10)

Average cell temperature at 3C discharge Current







Results & Discussion (9/10)

Surface temperature distribution (5X5 matrix) at the time of maximum average temperature for 75 Ah cell at 3 C discharge current with Al. cooling plate setup







- ➤ Average cell temperature can be decreased significantly with AI. cooling plates, but ΔT increases
- Maximum surface temperature can be observed near the cell tabs





- Scale up models can be used to study the performance of large format cells
- >∆T is influenced by ambient temperature, c-rate and cell size
- Improved thermal management should be the main focus for working with large format cells





- The cell model can be improvised by including more sub-models per unit area to simulate spatial inhomogeneity
- Aged cells should included in future studies





Thank you for your attention

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