

# Functional aspects of the X-ray structure of mitochondrial creatine kinase: A molecular physiology approach

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## Abstract

Mitochondrial creatine kinase (Mi-CK) is a central enzyme in energy metabolism of tissues with high and fluctuating energy requirements. In this review, recent progress in the functional and structural characterization of Mi-CK is summarized with special emphasis on the solved X-ray structure of chicken Mi<sub>v</sub>-CK octamer (Fritz-Wolf *et al.*, *Nature* 381, 341–345, 1996). The new results are discussed in a historical context and related to the characteristics of CK isoforms as known from a large number of biophysical and biochemical studies. Finally, two hypothetical functional aspects of the Mi-CK structure are proposed: (i) putative membrane binding motifs at the top and bottom faces of the octamer and (ii) a possible functional role of the central 20 Å channel. (*Mol Cell Biochem* **184**: 125–140, 1998)

**Key words:** active site, catalytic mechanism, creatine kinase, high-energy phosphate transport, membrane binding, metabolic channeling, mitochondrial energetics, octamer/dimer equilibrium, X-ray structure

**Abbreviations:** ANT – adenine nucleotide translocator; Cr – creatine; cCr – cyclocreatine; CK – creatine kinase; Mi-CK – mitochondrial CK isoforms; Mi<sub>v</sub>-CK – sarcomeric Mi-CK isoform; MM-CK, BB-CK – cytosolic CK isoforms; PCr – N-phosphoryl-creatine; SR – sarcoplasmic-endoplasmic reticulum; TSAC – transition state analogue complex

## Introduction

The ‘energy-rich’ compound, N-phosphoryl creatine (PCr) was first identified in 1928 in muscle [1] and was initially thought to represent the direct chemical energy source for muscle contraction. However, the discovery of the enzyme ATP:creatine phosphotransferase or creatine kinase (CK, EC 2.7.3.2) in 1934 by Lohman [2] showed that the N-phosphoryl group of PCr is transphosphorylated onto ADP to regenerate ATP [3], the latter being the universal energy currency in all living systems [4]. CK was first purified in 1954 by Kuby *et al.* [5] from rabbit muscle and its properties were described by Kuby and Noltman [6]. At this time and for many years to come, CK was considered a strictly soluble

enzyme simply fulfilling the role of a temporal ATP buffer or back-up system, as still depicted in current textbooks of biochemistry and cell biology.

However, after the discovery of (i) the cytosolic isoforms of CK [7], i.e. muscle-type MM-CK and brain-type BB-CK, (ii) the developmental transition of CK isoforms during muscle cell differentiation from ubiquitous BB-CK to sarcomeric muscle-specific MM-CK via the transitory MB-CK hybrid [8] and (iii) the identification of a novel mitochondrial CK isoform (Mi-CK), strictly confined to mitochondria [9], a different view of the enzyme’s function emerged. Right after the discovery of Mi-CK, it was postulated that ATP, formed by oxidative phosphorylation in muscle, would be transphosphorylated to PCr which then

would diffuse to the myofibrils and be converted into ATP again ([10], p.170 and [11, 12]). The proposal that creatine may stimulate mitochondrial respiration was born soon after [13] and this concept has been corroborated by numerous consecutive studies showing, for example, a functional coupling of Mi-CK with the adenine nucleotide translocator (ANT, for a review see [14]).

After the discovery of the isoenzyme-specific attachment of MM-CK to the myofibrillar M-band of sarcomeric muscle [15, 16], the idea gained support that the ATP-consuming contractile apparatus was connected to sites of energy production, e.g. mitochondria, via PCr and the CK isoforms located at these respective sites. Several models have since been proposed to describe the complex cellular energy supply system: The PCr-shuttle [17–23] or PCr-circuit [24–27] (for a collection of relevant publications and references see a special issue of *Mol Cell Biol* [28]). The idea behind the PCr-circuit model is that the PCr/CK system not only serves as a temporal energy buffer for ATP [29], but also as a spatial energy transport system connecting cellular sites of ATP-consumption, i.e. ATPases, to sites of energy production, i.e. glycolysis and/or oxidative phosphorylation. The latter function is especially important in highly polar cells such as photoreceptor cells or spermatozoa [30] where diffusion limitations of ATP and ADP can be overcome by PCr via the CK system [31].

In addition, the PCr-circuit represents a regulatory system for the physiologically required maintenance and adjustment of subcellular ATP/ADP ratios. For example, CK associated with the sarcoplasmic-endoplasmic reticulum (SR) [32] is thermodynamically important for efficient  $\text{Ca}^{2+}$ -pump activity, as shown both *in vitro* with isolated SR vesicles [33–35], as well as *in situ* with skinned muscle fibres [36]. Probably, the role of CK in the energetics of  $\text{Ca}^{2+}$ -homeostasis in excitable cells [30] represents the most crucial function of the CK system in general. This has recently become evident from analysis of the muscle phenotype of transgenic knock-out mice lacking the expression of both cytosolic MM-CK and sarcomeric  $\text{Mi}_b$ -CK. Skeletal muscles of these mice show problems with  $\text{Ca}^{2+}$ -release and uptake and display significantly prolonged relaxing times [37].

The temporal buffer function of the CK system, which seems generally accepted now (see above), has been corroborated elegantly by a transgenic approach with mice expressing BB-CK in the liver, where normally no or only very little CK activity is found. In the perfused livers of such transgenic animals, ATP levels and pH remain constant much longer after a fructose overload than in control livers [38]. In addition, the transgenic CK livers, if perfused with creatine, are protected by endogenously formed PCr from hypoxia and ischemia [39] and, due to a reinforced hepatic energy metabolism, regenerate significantly better, after 70% hepatectomy, compared to control livers [40].

The major point of discussion about the multiple functions of the CK system, at the present time, is related to the spatial buffer or transport function via PCr-shuttling. Results obtained by *in situ*  $^{31}\text{P}$ -NMR saturation transfer measurements investigating the CK-mediated reaction flux at different work loads of muscle [41] or under creatine depletion [42] have been interpreted as proof against the PCr shuttle. However, the unexpectedly anomalous behaviour in  $^{31}\text{P}$ -NMR terms of transgenic mice expressing graded CK levels [43] suggested that some of the CK-flux, measured *in situ* by saturation transfer  $^{31}\text{P}$ -NMR, may evade detection by this method [44]. This problem seems to be exacerbated by the fact that, due to compartmentation of adenine nucleotides *in vivo*, some ATP may be invisible by NMR as well [45], indicating that a fresh look at the NMR data is needed and that this problem can only be solved by a multidisciplinary approach. Within this context, the dynamic interplay between the CK and the adenylate kinase system is also of relevance [46].

The recently solved X-ray structure of octameric  $\text{Mi}_b$ -CK [47] sheds new light on some important functional aspects of the CK system discussed above and seems to be consistent with the proposed energy channeling function of this enzyme [48]. The new insight into the atomic structure of the  $\text{Mi}_b$ -CK molecule allows now for a detailed structure/function analysis concerning the molecular physiology of this enzyme, its catalytic site and mechanism, its octamer/dimer equilibrium, as well as its interaction with mitochondrial membranes and the relevant proteins thereof. Some of these new and intriguing aspects of the  $\text{Mi}_b$ -CK structure are described here and certain prominent key features of the molecule are discussed below within the physiological framework of mitochondrial and cellular bioenergetics.

The first part will present a short overview of the chicken  $\text{Mi}_b$ -CK structure in the context of our present biochemical and biophysical knowledge about CK isoforms, including active site residues and catalytic mechanism. In the second part, models will be developed, based on the  $\text{Mi}_b$ -CK structure, to elucidate the unknown structural basis of some  $\text{Mi}_b$ -CK-specific functions, e.g. the strong binding to mitochondrial membranes and the functional coupling and metabolite channeling to ANT of the inner mitochondrial membrane and porin (or voltage-dependent anion channel, VDAC) of the outer mitochondrial membrane. Two structural features and their possible functional consequences will be discussed in detail: (i) the identical fourfold top and bottom faces of the octamer which contain putative membrane binding motifs likely to be involved in binding to mitochondrial membranes and, possibly, interaction with ANT or porin and (ii) the central, 20 Å wide channel which may be of significance for the exchange of energy metabolites between mitochondria and cytosol under the assumption that  $\text{Mi}_b$ -CK follows a ‘back door’ mechanism.

### Crystal structure of chicken $Mi_b$ -CK

CK is one of the enzymes with the longest history in crystallographic studies. Crystallization attempts date back to the 1950s, crystallization of cytosolic [5, 49–54] and mitochondrial [55, 56] isoenzymes have been reported by different groups. Despite of all these results, the first successful solution of the crystal structure of a guanidino kinase could only be reported recently [47], using chicken  $Mi_b$ -CK crystallized in our group [55].

The reason for the successful crystallization may have been the choice of the animal species, i.e. the chicken, with a body temperature of 42°C, as well as the choice of the mitochondrial CK isoform ( $Mi_b$ -CK), which had never before been used in crystallization trials. Octameric chicken  $Mi_b$ -CK from sarcomeric muscle turned out to be remarkably stable against denaturation and was surprisingly insensitive towards proteolysis, so that enzymatically active protein, mostly octameric  $Mi_b$ -CK, could be recovered from crystallization setups even after months of storage at 25°C.

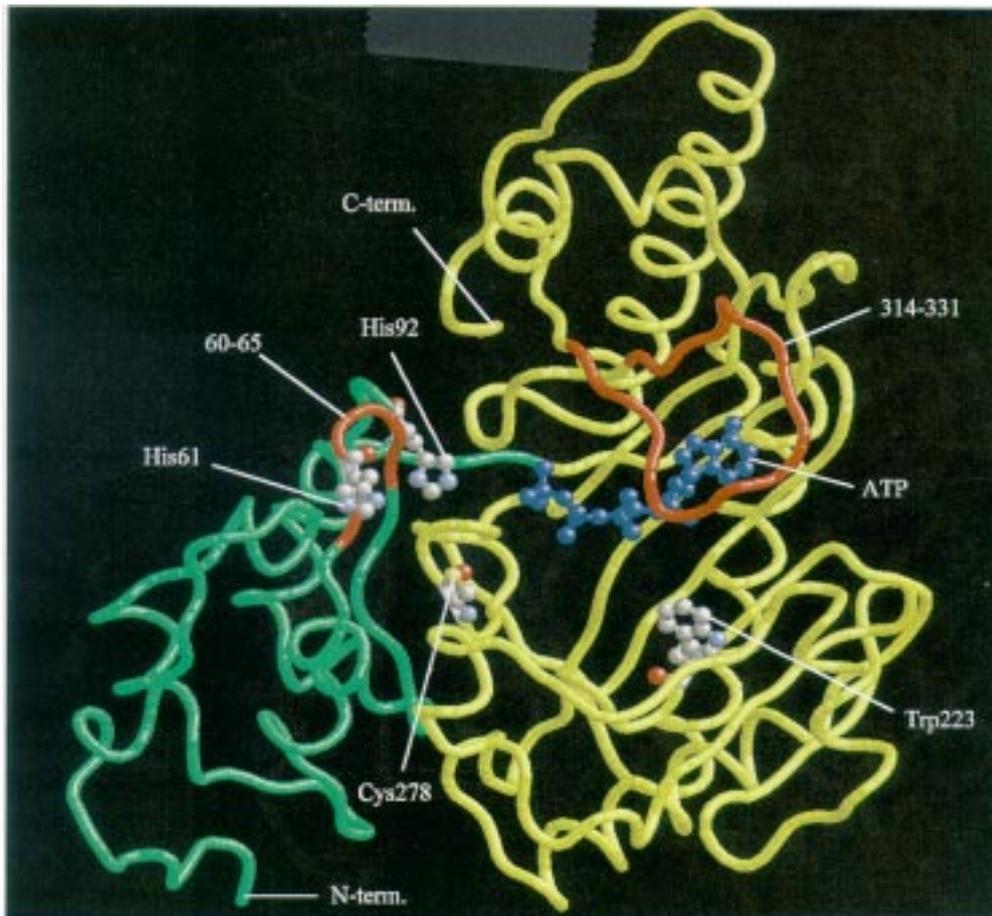
$Mi_b$ -CK was crystallized by the hanging drop approach using protein purified from chicken heart tissue [55, 56] or, later, from overexpressing *E. coli* bearing expression plasmid pRF23 [57]. Several tetragonal crystal forms were observed, two of which proved useful for crystallographic studies of the enzyme in the presence or absence of substrates (free enzyme and  $Mi_b$ -CK × ATP: space group  $P4_2$ ,  $a=126.0$  Å,  $c=144.7$  Å; free enzyme: space group  $P4_2$ ,  $a=171$  Å,  $c=150$  Å). The use of a single isomorphous derivative and a complex theoretical approach involving phase extension (W. Kabsch, unpublished) allowed the calculation of an electron density map to 3 Å resolution. It should be noted here that the intrinsic non-crystallographic symmetry of octameric  $Mi_b$ -CK was crucial for the solving the crystal structure. The polypeptide chain was fitted to the electron density map using the interactive program O [58] and the model refined in several steps using the X-PLOR software [59].

The course of the polypeptide chain of a monomer is shown in Fig. 1. The structures of  $Mi_b$ -CK-ATP and of the free enzyme were found to be very similar. Moreover, structural differences between the four crystallographically independent monomers in the octamer are rare, comprising residues 1–10, 60–66, 109–120 and 313–327. Uncertainties in the structure assignment comprise the five N-terminal residues and the loops 60–65 and 316–326 (see Fig. 1), which have a high temperature factor above 40 Å<sup>2</sup>. This factor is a displacement parameter for an atom in a crystal, containing input from disorder in atomic positions as well as from thermal vibrations and thus indicates flexibility inside a structure. The overall dimensions of the monomer are approximately 36 Å × 42 Å × 69 Å. Assignment of the secondary structure by the procedure DSSP [60] is given in Fig. 2. Of the 380 residues, 33% are  $\alpha$ -helical and only antiparallel  $\beta$ -strands were found.

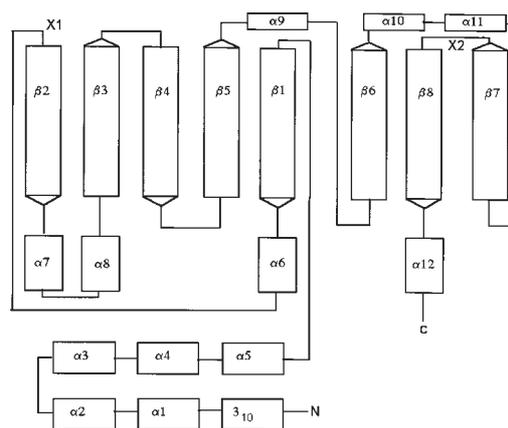
This is in good agreement with the results obtained by circular dichroism spectroscopy [61]. The monomer consists of a small (residues 1–112) N-terminal (domain I) and a large (residues 113–380) C-terminal domain (domain II) with the ATP binding site located in the cleft between the two domains (see Fig. 2). The small domain I is built from a short  $3_{10}$ -helix and five  $\alpha$ -helices of which helix  $\alpha_3$  is somewhat distorted. The large domain II contains an eightstranded antiparallel  $\beta$ -pleated sheet, flanked by seven  $\alpha$ -helices. As seen in Fig. 2, some of the strands in the large  $\beta$ -sheet are unusually connected. Moreover, the  $Mi_b$ -CK fold is different from all other kinases for which a three-dimensional structure has hitherto been determined, although there are some similarities in the fold of domain II to glutathione S-transferase (E. Di Iorio, ETH-Zürich, personal communication). Figure 2 also shows the unusual arrangement of the  $\beta$ -strands in the eight-stranded  $\beta$ -pleated sheet of domain II, which can also be considered to consist of two nearly equally sized subdomains IIa and IIb, separated by  $\beta$ -strand 1.

As CK is known to be remarkably stable towards proteolysis, it is interesting to note that the main nicking site of CK susceptible to proteases, Asp323 [62] (Fig. 2), is located on the highly flexible loop 316–326. After partial unfolding of muscle-type cytosolic M-CK by denaturants, another region around amino acids 158–166 becomes sensitive to staphylococcal proteinase V8 and therefore was postulated to represent a domain/domain boundary [63]. In  $Mi_b$ -CK, this region forms an extended stretch between the  $\alpha_6$  and the  $\beta_2$  regions as shown in Fig. 2 but cannot be considered a domain boundary. Recent experiments employing two genetically engineered fragments of  $Mi_b$ -CK [64], nevertheless, suggest that this region is probably defining the boundaries of folding entities within the  $Mi_b$ -CK structure since, more recently, a  $Mi_b$ -CK fragment consisting of residues 167–380 was shown to be an independent folding entity, whereas the fragment 1–166 could not fold to its native conformation (M. Forstner *et al.*, unpublished).

The octamer is held together by numerous interactions of its monomers.  $Mi_b$ -CK octamers are quite stable and the dissociation into dimers upon dilution takes hours to weeks [65]. This process is greatly accelerated by the addition of a transition state analogue complex of the substrates (TSAC = creatine, Mg-ADP and nitrate), which leads to dissociation within minutes [66]. The dimers of all CK isoforms are very stable and disintegrate into monomers only under chaotropic conditions. From this information it was possible to define the dimeric building blocks of the octamer from the number of interactions between the monomers, under the condition that the monomer-monomer interactions forming the dimer have to be more numerous and extensive than the interactions holding the dimers together to yield the octamer. The validity of this approach was supported by the finding that the  $Mi_b$ -CK dimer could be used to model the structure of dimeric M-



*Fig. 1.* Monomer structure and flexible regions of the  $Mi_b$ -CK monomer. The course of the polypeptide chain is shown in the 'backbone trace' representation as provided by the macromolecular rendering software RASMOL v. 2.6 [123]. Domain I is shown in green and domain II in yellow; the bound ATP is colored in blue. Flexible segments of the polypeptide chain with temperature factors above  $40 \text{ \AA}^2$  are colored in red. The position of the flexible loops (60–65 and 314–331, comprising the highly flexible stretch 316–326) and of the N- and C-termini is indicated. For His61, His92, Trp223 and Cys278, the side chains are drawn in a 'ball-and-stick' representation (C atoms in gray, N atoms in blue, O atoms in red, S atoms in yellow).



*Fig. 2.* Chain fold of the  $Mi_b$ -CK monomer. Schematic representation of secondary structure of  $Mi_b$ -CK. First and last amino acid residues in the helices and sheets were determined by the use of the automatic procedure DSSP [60]. The molecule can be partitioned into two domains. The small domain I consists of a  $3_{10}$  helix [P11–D14] and five  $\alpha$ -helices of which helix  $\alpha 3$  is distorted [ $\alpha 1$ :C24–C28;  $\alpha 2$ :P31–L37;  $\alpha 3$ :L48–D57;  $\alpha 4$ :S76–V79;  $\alpha 5$ :A81–R91]. The large  $\beta$ -domain contains a central  $\beta$ -stranded antiparallel  $\beta$ -sheet [ $\beta 1$ :V121–R130;  $\beta 2$ :G166–S170;  $\beta 3$ :G211–N215;  $\beta 4$ :F220–I224;  $\beta 5$ :T230–K237;  $\beta 6$ :R287–K293;  $\beta 7$ :L312–G316;  $\beta 8$ :V328–N333] surrounded by seven  $\alpha$ -helices [ $\alpha 6$ :R143–A158;  $\alpha 7$ :E176–D185;  $\alpha 8$ :P195–T198;  $\alpha 9$ :M241–R262;  $\alpha 10$ :P295–K299;  $\alpha 11$ :R302–L310;  $\alpha 12$ :E341–E363]. Two protease sensitive sites are indicated (X1: V8-protease sensitive site at amino acid residue 166; X2: multiple protease-sensitive site on the flexible loop 316–326 at amino acid residue Asp323, see text for details).

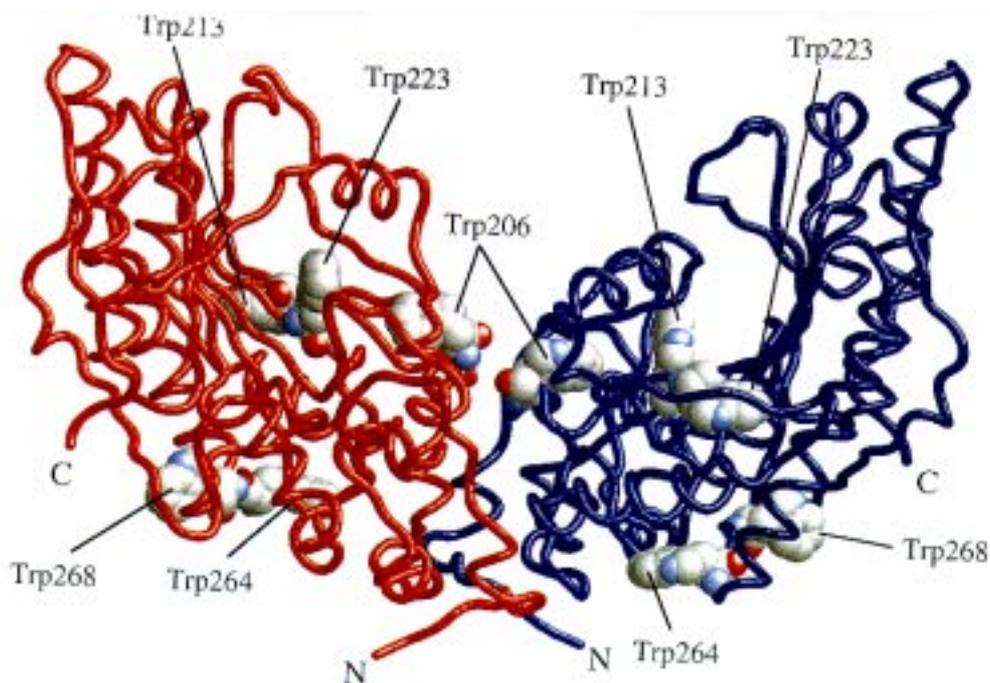


Fig. 3. The tryptophan residues of  $Mi_b$ -CK. A dimer of  $Mi_b$ -CK is shown in the 'backbone trace' representation as provided by the rendering program RASMOL v. 2.6 [123] with one monomer being colored red, the other blue. The side-chains of the tryptophan residues Trp206, Trp213, Trp223, Trp264, and Trp268 are shown in a 'spacefill' model representing the van der Waals-surface. The involvement of Trp206 in the monomer-monomer contact is evident from the figure, the position of Trp223 in the active site is also obvious and Trp264 is located at the dimer/dimer interphase (see Fig.3b in [47] and [67]). The N- and C-termini are indicated.

CK and compare the calculated scattering function of the model to the experimentally obtained small-angle scattering curve of M-CK in solution (Forstner *et al.*, submitted).

Moreover, the contribution of Trp206 (Fig. 3) in one of the monomer/monomer contact regions to the stability of the dimers has been proven by site-directed mutagenesis and

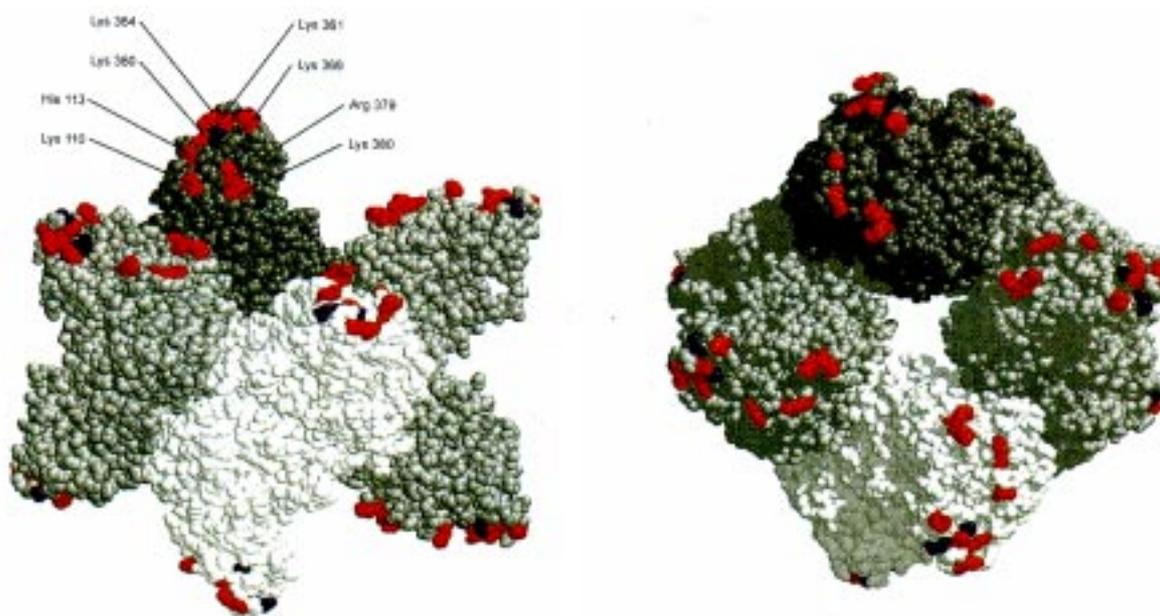


Fig. 4. Charged amino acids in the putative  $Mi$ -CK membrane binding motifs. The 'spacefill' model of the chicken  $Mi_b$ -CK octamer, with identical top and bottom faces for symmetry reasons, is shown (left) in a slightly tilted side view and (right) in a view onto the top (or bottom) membrane binding face (side length 96 Å) with the central 20 Å wide channel. Basic amino acids (positively charged) are depicted in red, acidic amino acids (negatively charged) are shown in blue. The four dimers forming the octameric structure are marked in different grey levels. The model representing the van der Waals-surface was prepared using the rendering software RASMOL v. 2.6 [123].

fluorescence spectroscopy [67] and the necessity for the presence of Trp264 [67], as well as of the intact N-terminus for octamer formulation has been demonstrated as well [68].

#### *Active site residues and catalytic mechanism*

Data analysis of the ATP containing co-crystals of  $Mi_b$ -CK reveals additional electron density for one nucleotide in each monomer which is not present in the native map. It is located in the cleft between the domains in the highly conserved region of all guanidino kinases. The adenine base fits into a pocket formed by the residues His186, Met235, Ser123, Arg125, His291, Gly289, and the pair Arg287, Asp330 which forms a salt bridge. The phosphate groups of ATP interact with four arginine residues at positions 125, 127, 287, and 315. Two additional residues, Arg91 and Arg336, are found at a distance of 5 Å from the  $\gamma$ -phosphate. This region, also including His92, leaves enough room to accommodate creatine in an orientation that would allow a direct in-line transfer of the phosphoryl group between the bound substrates.

By a number of experiments with chemical modification of SH-groups using different SH-reagents in the presence or absence of substrates, or the affinity label epoxycreatine [69], Cys278 was identified as the highly reactive 'essential' cysteine residue (see also Fig. 1). However, further studies of this type yielded contradictory results concerning the essentiality of Cys278 for enzyme catalysis (see [70]). Nevertheless, until recently, Cys278 has been suggested in the literature as an important residue for catalysis [71]. Thorough kinetic analysis of several replacement mutants of Cys278 has solved this long debated issue by providing clear-cut evidence that Cys278 is not essential for catalysis per se, but is important for the synergism of substrate binding [72].

NMR studies on the role of histidines at the active site have revealed four such residues at distances of 12, 12, 14 and >18 Å from the  $Cr^{3+}$  in the paramagnetic  $\beta,\gamma$ -bidentate  $Cr^{3+}$ -ATP complex bound to the enzyme [73]. One of these histidine residues, with an altered intrinsic pK near 7, was postulated to act as an acid-base catalyst [74]. We find His92, His291, His229, His61 and His186 at distances of 12, 12, 14, 17, and 18 Å, respectively, from the O7 atom of ATP (see Fig. 1). Recent studies by our [75] and other groups [76] have shown, however, that none of these residues can be considered essential for the functioning of CK, although some of them certainly are of importance for substrate binding in the active site of CK. However, replacement mutants of His61 had a strong negative effect on the enzyme activity, e.g. the respective His to Ala mutant showed approximately 1 and 10% of wild type activity in the forward and backward reaction, respectively [75]. The importance of His61 for the catalytic activity of CK is consistent with the assumption that the flexible loop formed by residues 60–65 and which

is found rather distant from the ATP in the crystal structure might move towards the active site to promote catalysis [75].

Spectroscopic studies of cytosolic M-CK and its complexes with adenosine phosphates, as well as quenching studies, were suggestive for the occurrence of a tryptophan residue at or near the cosubstrate binding site [77]. Specific chemical modification of tryptophan residues revealed one of them as essential for enzymatic activity. These results are in agreement with an earlier  $^1H$ -NMR study indicating an interaction between protons of the adenine ring of bound ADP and one or more aromatic side chains of the protein. Trp206 and Trp223 are conserved throughout the whole family of guanidino kinases [78] and replacement of Trp223 resulted in a complete inactivation of the enzyme, indicating that Trp223 is situated near the active site and is crucial for catalytic activity and, therefore, was termed the 'active site tryptophan' [67]. By NMR, the active site tryptophan was found to be in the vicinity (< 5 Å) of the nucleotide substrate [78a].

Incidentally, four of the five tryptophan residues of  $Mi_b$ -CK (Trp206, Trp213, Trp223, Trp264, see Fig. 3) are found rather closely clustered and oriented in such a way that energy transfer, indicated by fluorescence studies [67] may indeed be possible, whereas Trp268 is closer to the C-terminal membrane interaction face of the octamer. In the X-ray structure of the wildtype enzyme, Trp223 was found at a distance of about 10 Å from the adenine ring. We assume, however, that in the presence of all substrates or at the transition state, the nucleotide would move 'down' towards Trp223, thereby bringing the substrate closer to this residue, thus explaining the distance measured by NMR. In this context it is necessary to remember that the Mg-nucleotide rather than the free nucleotide is the substrate for CK and that only the Mg-nucleotide can induce structural changes essential for catalysis [79].

In comparison with other kinase structures,  $Mi_b$ -CK has acquired a different folding pattern for catalyzing phosphoryl transfer, a result which is of significance for the evolutionary positioning of CK. Furthermore, Mi-CK is the only isoform in the creatine kinase system which forms octamers, and the structure supports the hypothesis that Mi-CK has evolved to serve an additional function as a structural protein that mediates the adhesion between inner and outer mitochondrial membranes and between cristae membranes via electrostatic interactions due to charged residues located at its four-fold faces [25–27]. These properties of Mi-CK will be discussed in detail in the next chapters.

#### *Membrane binding of Mi-CK*

In view of the physiological function of CK, one of the most important properties of this enzyme is the distinct subcellular

localization of its different isoforms (for a review see [25, 26]). This isoenzyme-specific localization has been related to the functional coupling of CK to energy providing and energy consuming processes of distinct subcellular sites and is part of the PCr-shuttle or -circuit model of CK function (see above). Mitochondrial CK isoforms, sarcomeric  $Mi_b$ -CK and ubiquitous  $Mi_a$ -CK differ from their cytosolic counterparts not only by their organellar compartmentation, but also by their membrane-binding property. While cytosolic CKs are mainly soluble but certain fractions of the enzyme are also bound to particular subcellular structures (e.g. myofibrils, sarcoplasmic reticulum and plasma membrane), Mi-CKs are known to bind strongly to the outer surface of the inner mitochondrial membrane [80] and, in case of  $Mi_b$ -CK, to bridge mitochondrial membranes (for reviews see [81] and Stachowiak *et al.*, this issue). Membrane binding of Mi-CK is considered the structural basis of functional coupling between Mi-CK and oxidative phosphorylation which occurs via the metabolic channeling between Mi-CK and the adenine nucleotide translocator (ANT) of the inner mitochondrial membrane (e.g. [82], see also below). Similarly, a functional interaction between the outer mitochondrial membrane porin and Mi-CK has been proposed [25].

It is generally accepted now that Mi-CK binds to membranes mainly by electrostatic interactions, although the participation of hydrophobic interactions cannot be ruled out [83]. As binding partners in the mitochondrial membranes, cardiolipin and other phospholipids were identified by different methods [83–87]. Highest affinity was determined for cardiolipin which contains a two-fold negatively charged headgroup and occurs mainly in the inner mitochondrial membrane. However, it is still debated whether the octamer only binds to the lipid surface [88] or penetrates partially into the membrane bilayer [87]; the latter interaction would be favored by cardiolipin which is capable to form non-bilayer structures [89, 90]. Such a ‘dipping-in’ could also explain the resistance of a certain Mi-CK fraction against detachment by high ionic strength or detergents and the arrangement of the bulky enzyme in the rather narrow mitochondrial intermembrane space.

Although an interaction between Mi-CK and ANT or porin has been shown on the functional level (for a review see [25]), direct evidence for a structural interaction *in vivo*, e.g. by crosslinking studies, has not been obtained so far [81, 91]. Furthermore, reconstituted ANT did not bind to Mi-CK [85] and electrostatic repulsion of the basic Mi-CK isoforms is suggested by the basic pI of ANT [84] and positive charges exposed to the intermembrane space by porin [92]. However, *in vitro* complex formation between Mi-CK, ANT and porin has clearly been demonstrated [93–95]. Since ANT is situated in a cardiolipin membrane patch [96, 97], the binding of Mi-CK to such patches is assumed to bring the ANT and Mi-CK in close vicinity to each other [25, 98, 99]. The observed interaction of Mi-CK with porin might occur by a similar

complex formation via charged phospholipids [81] or by a different, hydrophobic binding mode [94].

#### *Early studies on Mi-CK membrane binding domains*

Mi-CK-membrane interactions have been studied mainly with respect to binding mode, binding constants and possible binding partners of Mi-CK; little is known about the structural domains of Mi-CK involved in membrane binding. In a first attempt to identify membrane binding domains, peptide fragments of rat sarcomeric  $Mi_b$ -CK, obtained by cyanogen bromide digestion, were analyzed in a binding assay with artificial cardiolipin vesicles [98]. Only one peptide, containing the 25 N-terminal amino acids, showed high affinity to cardiolipin. This approach, however, is prone to artifacts since an isolated peptide can behave differently compared to the same amino acid sequence in its native protein environment (see also below). To identify amino acids involved in membrane binding, the intact Mi-CK molecule was subjected to chemical modification of selected amino acids (Met, Cys, His, Lys and Arg) [98]. It occurred that, while a modification at Met, Cys, or His residues still allowed binding to cardiolipin, modification of Lys or Arg residues drastically reduced binding by over 90%. Although these results must be interpreted with caution, since chemical modifications could introduce overall structural changes, they corroborate evidence on electrostatic interaction between basic amino acids of Mi-CK with negatively charged cardiolipin.

A different early approach for the structural characterization of Mi-CK and its membrane interaction sites applied a variety of electron microscopic techniques for single molecule imaging [100–102]. These methods visualized for the first time the octamer structure of Mi-CK, conserved throughout different species from sea urchins to mammals [103], thus confirming ample biochemical and biophysical evidence on the octameric structure of membrane-bound Mi-CK *in vitro* (e.g. [65, 102]) and *in vivo* (e.g. [104]). In addition, these studies revealed the fourfold symmetry of the octamer with dimers arranged along the symmetry axis (compare to the atomic structure in Fig. 4). Most importantly, they showed that top and bottom faces of the octamer are identical and confer the ‘sticky’ character to the protein which is responsible for binding to artificial cardiolipin-rich membranes and other negatively charged surfaces [100–102]. Only under very particular conditions in solution, due to the absence of surfaces, top and bottom faces attach to each other and form Mi-CK filaments [101].

#### *Putative membrane-binding motifs of Mi-CK*

So far, information on the structural basis of Mi-CK membrane binding has been scarce. The recently solved octameric



(Lys360, Lys361, Lys364, Lys369, Lys380) and one arginine (Arg379). Out of the four negative charges (Asp357, Glu359, Glu363, Asp367), only the two asparagines (Asp357 and Asp367) are pointing to the binding face (marked in blue in Fig. 4). The homologous stretches of cytosolic CK isoforms (B-CK and M-CK) do not show such a density of positive charges. It has to be mentioned that the C-terminal sequence is not absolutely identical between sarcomeric and ubiquitous Mi-CK. While the number of basic amino acids is well conserved, their position and a sequence stretch at the very C-terminus show differences which could be responsible for divergent membrane binding properties of the two Mi-CK isoforms (Fig. 5). In both isoenzymes, charged residues at the C-terminus (Arg379 and Lys380) are separated from the other charges (Lys 360, Lys361, Lys364, Lys369) by a stretch of 8–9 residues containing 6–7 hydrophobic amino acids (Fig. 4). These stretches may possibly enter a phospholipid bilayer and thus contribute to the strengthening of the Mi-CK-membrane interaction.

The second sequence motif, a short internal amino acid stretch (Ala107 - Gln115) located at the interconnection between helices of the small and large Mi-CK domain [47] harbors one to three positive charges depending on the Mi-CK isoform and the species considered. In chicken Mi<sub>b</sub>-CK, one lysine (Lys110) and one histidine (His113) are found (Fig. 4). By contrast, B-CK does not contain these positive charges and the single positive charge in M-CK is surrounded by several negative charges (Fig. 5). In all Mi-CKs, the cumulative positive net charge of both motifs is about four and is mainly due to lysines (Fig. 4). A similar result, i.e. four lysines involved in Mi-CK-membrane binding, was obtained by a pure modeling approach based on the pH-dependence of the binding process [105].

Although the basic amino acids mentioned are not contiguous on the Mi-CK primary structure, their arrangement on the octamer surface is such that they form clusters of two or four positive charges (Fig. 4). This could have a special significance, given the proximity of the two negatively charged phosphates of the cardiolipin headgroup. A similar ‘collar’ of lysines has been identified in ANT and proposed to mediate the contact to cardiolipin [106]. Another common feature of both putative membrane-binding motifs is the relatively high temperature factor compared to the main part of the Mi-CK structure [47]. The flexibility indicated by this fact may be useful for a ‘docking’ onto the negatively charged phospholipid heads.

Taken together, the properties of the two sequence motifs discussed fully support their putative function in membrane-binding, provided that the Mi-CK is in the octameric form and membrane binding occurs primarily due to electrostatic interactions with negatively charged membrane phospholipids and, possibly, a minor participation of hydrophobic forces. This binding model could explain the ability of the Mi-CK

octamer to bridge two membranes *in vitro* by its top and bottom faces and to interact functionally with ANT and porin *in vivo* by a close co-location with these membrane proteins inside cardiolipin/phospholipid membrane patches. Formation of the proposed multienzyme complex for energy export (ANT, Mi-CK and porin) could be a transient and dynamic process, similar to the formation of contact sites. A detailed site-directed mutagenesis study of the putative membrane binding motifs of Mi-CK is underway and should provide a clue for defining the interaction sites at the Mi-CK-membrane interface.

#### *Metabolic channeling - a mechanism for Mi-CK in mitochondrial contact-sites?*

One of the most important requirements of a cell is to save metabolic energy and to control the flux of metabolites for the biosynthesis of compounds necessary to keep the plethora of biochemical reactions going. These requirements are of particular importance for reactions which directly influence the metabolic status of a cell and its future fate, e.g. energy-conversion reactions like the turnover of ATP, ADP, PCr and Cr catalyzed by CK isoforms. One of the mechanisms that evolved to reach a maximum of energetic efficiency and metabolic control is metabolic channeling: a precursor which is synthesized by enzyme (or subunit) A is passed to enzyme (or subunit) B without getting into contact with the bulk solution (for a review see [107]). Metabolic channeling has been well described for tryptophan synthase which possesses a 25 Å hydrophobic tunnel linking the active sites of two different subunits [108] and thus passing the substrate indole from the  $\alpha$ - to the  $\beta$ -subunit. However, channeling can occur between huge covalently linked enzyme-complexes such as the fatty acid synthase, as well as between the structurally less tight coupled glycolytic enzymes glyceraldehyde-3-phosphate dehydrogenase and aldolase [109].

The sequestering of intermediate products in a micro-compartment provides several advantages for the reaction sequence, but also for cellular metabolism in general. Since these intermediate products immediately serve as substrates for the subsequent reaction and do not equilibrate between microcompartment and bulk solution, which often would favor the backward-reaction, the difference in Gibbs’ free energy for their formation is decreased. Competing side-reactions may be excluded and counteracting catabolic and anabolic pathways are separated thus preventing interference which would perturb metabolism. Furthermore, sequestered intermediates are present at high local concentrations and an apparently low  $K_m$  for these intermediates can be observed with the channeling complex compared to the non-channeling situation, e.g. *in vitro* measurement with isolated components. Finally, channeling intermediate products of a reaction

cascade in a microcompartment also allows to strictly control the whole process by feed-back regulatory mechanisms such as substrate activation, product inhibition and cooperativity and the over-all-flux of the reactions can be controlled very accurately, e.g. by a limiting step in the reaction sequence (for an overview and further information on the topic see 'Cell Architecture and Metabolic Channeling' by Ovàdi [107] and 'Channeling in Intermediary Metabolism' by Agius and Sherratt [110]).

As already mentioned before, there is functional as well as *in vitro* evidence for complexes containing Mi-CK, porin and ANT, located in the contact sites of the mitochondrial envelope. An important function of these ANT/Mi-CK/porin complexes could be metabolite channeling [48]. In contrast to tryptophan synthase, the function of a Mi-CK complex would not be a reaction cascade where an intermediate is transferred, but the transport of energy-rich phosphate compounds like ATP, ADP and PCr between subcellular compartments. This means that ATP reaching the mitochondrial intermembrane space via ANT is directly channeled to the Mi-CK octamer and used for synthesis of PCr which then is channeled out of the mitochondrion into the cytosol via Mi-CK and porin. Vice versa, ADP is back-channeled into the matrix via ANT.

A prerequisite for this mode of action is a physical interaction of Mi-CK and ANT or, at least, a close vicinity of both leading to 'leaky' channeling. The finding that externally added pyruvate kinase is not able to use the ADP produced by the Mi-CK reaction [82, 111] and the fact that about 50% of Mi-CK activity is resistant against iodoacetate treatment [93] strongly supports the model of Mi-CK in close vicinity to ANT and porin. Kottke and co-workers further showed that in mitoplasts supplied with ATP and creatine, the blockage of porin by a polyanion strikingly decreases Mi-CK activity [93]. In addition, creatine-stimulated mitochondrial respiration due to ADP synthesis via Mi-CK is not abolished by the addition of external ADP traps like PEP plus pyruvate kinase, showing that the ADP generated by Mi-CK is located in a microcompartment close to the inner mitochondrial membrane and does not equilibrate with the bulk ADP pool [82, 111]. An interaction of Mi-CK with porin is indicated by the fact that isolated porin is able to induce octamer formation of an N-terminal Mi-CK deletion mutant which is unable to build up octamers in free solution [94]. Finally, complex formation of octameric Mi-CK with porin and ANT has recently been demonstrated *in vitro* [94, 95].

There are several advantages of the porin/Mi-CK/ANT arrangement: (i) ATP originating from the matrix space does not equilibrate with the cytosolic ATP and is present at very high local concentrations which facilitate the formation of PCr from Cr; (ii) since ATP is directly turned over after being exported out of the matrix, it will generate high local concentrations of ADP in the intermembrane space which stimulates respiration and favors re-import of ADP into the matrix

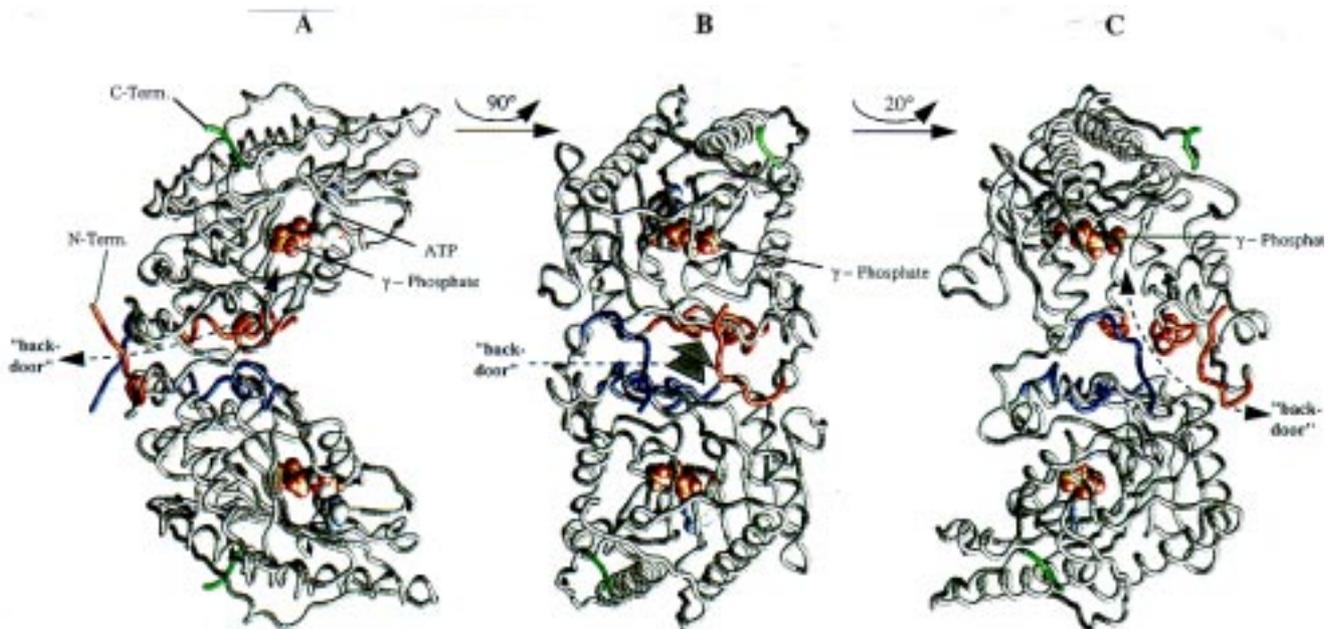
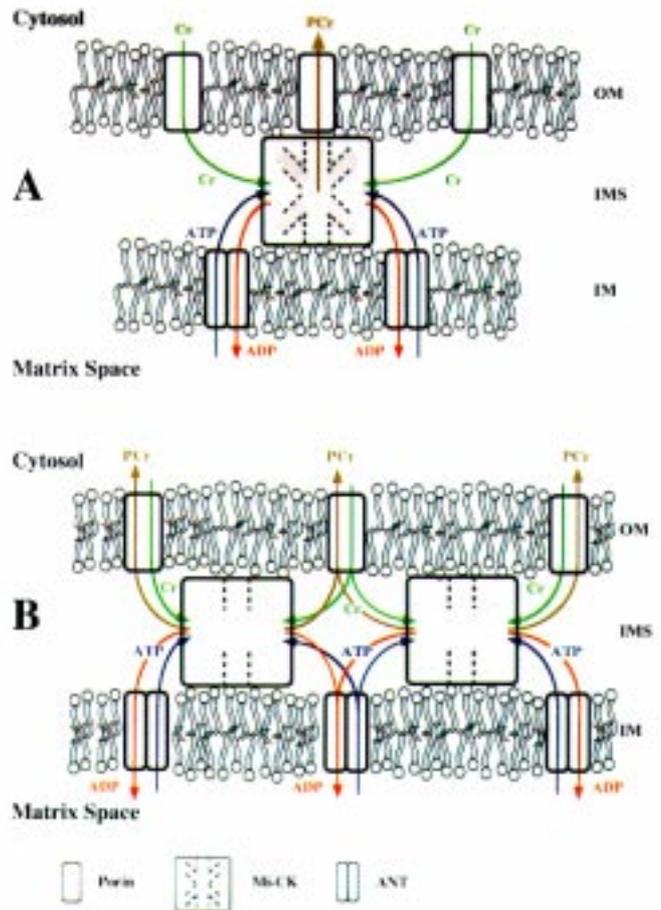
because of a steeper ADP gradient across the inner membrane [111]. This model also allows the ANT to act as an obligatory antiporter, since for every exported ATP there is an ADP molecule present which can be bound by the translocase. The thermodynamic advantage of such a complex is due to (i) the removal of the translocation product ATP, favoring further export of the latter via ANT, and (ii) the removal of once formed PCr from equilibrium by direct channeling into the cytosol via porin, favoring PCr synthesis by Mi-CK [111, 112]. Possible transport and diffusion pathways involved of such a channeling mechanism are discussed in detail in the next chapter (see also Fig. 6).

#### *Metabolite access to Mi-CK - a 'back door' hypothesis*

As it has been outlined above, Mi-CK displays functional and possibly also structural coupling to ANT and porin, two membrane proteins involved in the transport of Mi-CK substrates and products. If this coupling involves metabolic channeling, a vectorial transport of substrates and products would be expected [48]. The X-ray structure of chicken  $Mi_b$ -CK [47] provides a new and suitable tool to develop models for metabolite transport in the multiprotein complex bridging the two mitochondrial membranes as suggested from *in vitro* studies [93–95]. In this respect, a specific structural feature of the octameric structure is of particular interest, i.e. the central pore or channel of approximately 20 Å in diameter penetrating the octamer. This channel has already been detected by electron microscopic studies of chicken  $Mi_b$ -CK [101] and has been proposed to connect the active site of Mi-CK with the surroundings, supposing that active sites were exposed towards the central channel. This assumption was plausible at a time where still no three-dimensional high-resolution structure of CK was available.

However, the X-ray structure of  $Mi_b$ -CK shows that the most obvious and direct metabolite access to the active site of Mi-CK is from the 'outside' of the molecule (see Fig. 3c in [47]) and not from the channel. The CK active sites, one for each monomer, are exposed on the concave side of the 'banana-shaped' dimers (Fig. 3), and four such dimers form an octamer by contacting each other via the convex 'backside' (involving an N-terminal stretch plus the region around Trp264, see Fig. 3 and also Fig. 3b in [47]). If the Mi-CK octamer binds to the inner and outer mitochondrial membranes by its identical top and bottom faces, as shown by Schnyder [102], the active sites face the intermembrane space surrounding the octamer and the channel is oriented orthogonally to the membranes and thus is excluded from direct interaction with the intermembrane space. For this kind of organization between Mi-CK and membranes, one could propose two different models which would explain a compartmentalized CK reaction as indicated by the functional coupling of Mi-CK to ANT and porin (see Fig. 6).

*Fig. 6.* Model considerations for microcompartmentation of the Mi-CK reaction. Two different models for microcompartmentation of substrates and products of the creatine kinase (CK) reaction, as well as for vectorial export of phosphocreatine out of the mitochondrion are shown. Both involve porin of the outer mitochondrial membrane (OM), adenine nucleotide translocator (ANT) of the inner membrane (IM) and Mi-CK octamers in the inter-membrane space (IMS). The fluxes of the metabolites are depicted by arrows colored in red (ADP), blue (ATP), green (Cr) and brown (PCr). (A): Possible 'back door' mechanism. Cr and ATP enter the CK active sites from the IMS. After phosphoryltransfer, PCr leaves the active site via the suggested 'back door' of the enzyme, being guided into the central 20Å wide channel of the Mi-CK octamer. From there, it is exported into the cytosol via porin situated on top of the central channel (see also text). This model suggests that the central hole in the octamer and porin indeed serve as a channel connecting the CK active site with the cytosol. (B): Microcompartmentation without the need for a 'back door' mechanism. Reactants and products of the CK reaction enter and leave the enzyme from the 'front door'. Octamers situated close to each other are suggested to form a microcompartment including porin and ANT for the transport of the metabolites to and from the cytosol or mitochondrial matrix, respectively. In this model, the central hole of Mi-CK has no channeling function and would serve merely a structural role. The vectorial transport of PCr to the cytosol would be regulated by the accumulation of the product in the IMS, resulting in a change of ion selectivity of porin (see also text).



*Fig. 7.* Structural model of an alternative leaving-route for the phosphocreatine product through a 'back-door' exit. We propose a possible pathway of phosphocreatine (PCr) through an exit between two CK-monomers leading to the central main channel of the octamer. A 'backbone trace' representation of a 'banana-shaped' Mi-CK-dimer is shown with bound ATP (A). The dimer is rotated about 90° (B) and additional 20° (C) around its longitudinal axis to provide a better 3-dimensional visualization. The C-termini are marked in green, the N-termini and residues 47–59/194–202 are shown in red (monomer 1) and blue (monomer 2). The alternative leaving-route for PCr is depicted in each representation with a dashed arrow; the broad arrow in (B) indicates the 'back door'. Note that this 'back door' is deduced only from the X-ray structure of Mi-CK and is postulated on the basis of biochemical considerations (see text). The figure was prepared using the rendering software RASMOL v. 2.6 [123].

In the first model, depicted in Fig. 6A, access of substrates and products to the active site occurs from different sides of the enzyme, that is from the intermembrane space and from the central channel. In this case, a role of the channel in the transport of metabolites would require (i) direct structural interaction with porin (and/or ANT) and (ii) accessibility of the channel from the active sites of CK, positioned on the outside of the four side-faces of the MiCK octamer, by a so called 'back door' pathway. The existence of an alternative route for substrate, product or water ('back door' mechanism) has been proposed recently for myosin [113], acetylcholine esterase [114] and the tryptophan synthase  $\alpha_2\beta_2$  multienzyme complex [115]. Such a 'back door' connecting the active site with the channel could, indeed, exist in Mi-CK, as shown in Fig. 7 for a Mi-CK dimer. Two helices in each of the monomers (residues 47–59 in domain I and 194–202 in domain II) and the N-termini form an opening face to the central channel and a small access route to the active site. However, this pathway would be probably too narrow for adenine nucleotides having a relatively large size and bulky conformation. Therefore, access of ATP and exit of ADP to and from the active site, respectively, is assumed to occur via the intermembrane space. It then follows that ANT of the inner mitochondrial membrane should be located in the vicinity of CK octamers without direct contact to the central channel, thus releasing ATP into the intermembrane space which would subsequently diffuse into the active site pocket of CK (Fig. 6A).

The described 'back door' pathway, however, would be best suited for PCr, since (i) the 'back door' directly points to a putative PCr/Cr binding site (facing the  $\gamma$ -phosphate of ATP) and (ii) the PCr molecule has a more elongated structure which could best fit through the 'back door' (due to the repulsion of charges, PCr becomes more elongated than Cr, which has a more or less cyclic structure). Thus, the model suggests that PCr is dissociating from the active site directly into the central channel of the octamer, thereby separating from ADP which cannot follow due to its larger size and also for electrostatic reasons. In addition, it is entirely conceivable that domain movement upon substrate binding leads to further opening of the 'back door', thus favoring PCr product release. In this model, the porin in the outer mitochondrial membrane would be in direct interaction with the central channel to transport synthesized PCr into the cytosol (Fig. 6A).

Although it is tempting to believe in the existence of such a 'back door' that would enable vectorial transport of PCr out of the mitochondrion or facilitate the proposed microcompartmentation of the CK reaction and its interaction with the transport proteins, alternative possibilities as the one depicted in Fig. 6B may be envisaged as well. In this model, several CK octamers arranged in close proximity (only two are drawn due to the limitation of the two-dimensional representation) enclose a part of the intermembrane space, thereby forming a microcompartment including ANT and porin. The central

channel of the Mi-CK octamer would then have no other function than a mere structural role, stabilizing the convex 'backsides' (see Fig. 3) of the four dimers in the octamer and probably offering space for structural rearrangements of the monomers within the octamer during catalysis, as has been shown to take place upon  $Mg^{2+}$ -nucleotide binding [79].

Several arguments, however, are in favor of a 'back door' mechanism (Fig. 6A). One of the strongest is the great thermodynamic advantage of such a mode of reaction due to the continuous withdrawing of one of the products from the active site environment. This mechanism would facilitate a non-equilibrium system, compared to the alternative model which releases all Mi-CK products into the microcompartment of the incoming substrates (Fig. 6B). Furthermore, it has been shown that product release is the rate limiting step of the CK reaction in the forward direction [74] and that PCr is the 'sticky' product. This would be in line with the certainly longer time for the diffusion of PCr out of the active site through a 'back door' compared to the fast release of ADP through the large open cleft in the CK structure. Another supportive argument comes from the inhibition mode of cyclocreatine (cCr). cCr, a competitive inhibitor of the CK reaction, has been shown to bind to the active site and to be phosphorylated [116]. The 'back door' model offers a simple explanation for this inhibitory mechanism. Although the cCr structure mimics the cyclic nature of Cr and cCr can be phosphorylated, the conformational change upon phosphorylation leading to a more elongated structure (see above) cannot take place like in the case of PCr. Therefore, phosphorylated cCr could be unable to use the release pathway of PCr through the 'back door' and thus would block the active site of Mi-CK.

The main argument against a 'back door' mechanism comes from the classical enzymological studies of the CK reaction. The forward reaction of CK was described to follow a random bi-bi mechanism [6, 74], while the proposed 'back door' pathway would rather imply a sequential scheme, involving binding of the substrates, domain closure induced by  $Mg$ -ATP [79] together with a possible further opening of the 'back door', PCr release via the 'back door' and, finally,  $Mg$ -ADP release into the intermembrane space. However, this mechanism is not necessarily in contradiction with the available kinetic data, since all of them have been obtained with dimeric MM-CK in free solution. Enzyme kinetics of Mi-CK might therefore be different, especially under *in vivo* conditions in the intermembrane space when attached to mitochondrial membranes. As of yet, there is no direct experimental proof for the proposed 'back door' mechanism in Mi-CK, and it should be noted once more that the appealing hypothesis presented herein is mere speculation based on structural data. Clearly, more studies addressing this intriguing question are necessary, including detailed enzyme kinetics of octameric Mi-CK, as well as a direct analysis of the 'back door' pathway, e.g. by introducing bulky residues with site directed mutagenesis.

## Conclusions and outlook

It has taken more than 60 years since the first description of CK [2] until the first X-ray structure of a member of the CK isoform family could be solved [47]. This step forward has been recognized as a significant achievement and landmark in CK research [71, 117]. As outlined above, the octameric structure has already provided some important insights and stimulating clues into the functioning of the enzyme and, after all, is also rewarding from a purely aesthetical point of view. However, having the structure of the Mi-CK octamer at hand with many details becoming obvious now, we are brought back to real life again. Some difficult longstanding questions, e.g. about the kinetic mechanism of the enzyme and the amino acid residues involved in catalysis, have still to be answered in detail. Are there true catalytic residues or does CK belong to the category of ‘cozymes’ designed to merely bring the substrates into close alignment [71]?

To solve this puzzle, an X-ray structure of CK in the presence of reagents known to induce the enzymes’ transition state, e.g. Mg-ADP, creatine and nitrate [118], will be of utmost importance. Such a structure of the ‘closed configuration’ of the enzyme should reveal not only the exact molecular outline of the active site pocket of CK caught in the act of catalysis, but should also shed some light on the expected hinge bending and domain movements taking place upon binding of Mg<sup>2+</sup>-nucleotide [79]. After having learned that each subunit of CK has its own catalytic site, a further important task will be the elucidation of possible cooperativity between monomers in the dimer and, even more interestingly, in the octamer. The latter question, of course, will be most relevant to the issues of vectorial metabolite channeling and directed export of PCr, derived by Mi-CK from matrix generated ATP, out of the mitochondrion (see above).

The understanding of the molecular details of CK catalysis and the possible design of CK-specific inhibitors are likely to be of great clinical relevance, since an involvement of CK in signal transduction [119, 120], as well as in tumor growth and malignancy has been demonstrated and since certain creatine analogues were shown to display anti-cancer activity on explanted human tumors [121] or to enhance significantly the effect of chemotherapy on cancer cells [122].

The precise knowledge of CK structure and function at the molecular level may elucidate the *in vivo* function of this isoenzyme system in cells and organs with high and fluctuating energy requirements and thus may complement the work done with transgenic mice lacking the expression of one or several CK isoform *in vivo* [37]. Since every novel result produces a myriad of new questions, much of what has to be learned is still ahead of us.

## Note added in proof

Recent <sup>31</sup>P-NMR studies comparing CK-mediated fluxes of inactive versus actively swimming sea-urchin sperm [124], as well as mathematical modeling of cardiomyocyte energy metabolism, including the known CK-‘knock-out’ models [125], are in support of a CK/PCr-shuttle in these cells.

With the help of the chicken Mi<sub>b</sub>-CK coordinates, the X-ray structures of cytosolic B-CK (Rao JKM and Wlodawer A, personal communication) and the related lobster arginine kinase (Dumas C and Janin J, personal communication) have been solved. Arginine kinase from horseshoe crab has been crystallized in the presence of TSAC substrates [126] and may provide the first transition state structure of a guanidino kinase. A new study of Mi-CK membrane binding properties [127] suggested a partial disorganization of the used phospholipid bilayer upon interaction with Mi-CK and revealed a decreased Mi-CK binding after treatment with p-hydroxymercuribenzoate. The latter may be due to a modification of the C-terminal Cys358 (see Fig. 5), which could influence the binding capacity of the C-terminus.

## Acknowledgments

Dr. W. Kabsch is gratefully acknowledged for discussion. This work was supported by the Swiss National Science Foundation (SNF grant No. 31-33907.92, to T.W.), by ETH graduate training grants for M.F., M.E. and O.S. and by financial support from the Swiss Society for Muscle Diseases, as well as by private sponsoring from Careal Holding AG and Synergen AG, Switzerland.

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